## Harikrishna Yadav. Nanganuru, Narasimhulu Korrapati / International Journal of Engineering Research and Applications (IJERA) ISSN: 2248-9622

www.ijera.com Vol. 2, Issue 3, May-Jun 2012, pp.2217-2219

# STUDIES ON BIOSORPTION OF CADMIUM BY Pseudomonas putida

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#### **ABSTRACT**

Industrial waste and sewage are serious and growing problems in most developing countries. Some heavy metals are known to cause damage to living organisms including human beings. Their detoxification efficiency indicates good potential for application in bioremediation of toxic heavy metals. In those heavy metals, Cadmium in soil is influenced to a great extent by microbial activity. Microorganisms have ability to absorb Cd from a medium. The biosorption removal of cadmium ions from aqueous solutions by using the bacterial culture was investigated. The growth of the Pseudomonas putida culture was noticed by Cd concentration in growth medium, thus about 20% of the bacteria can grow up to 40 mg Cd/L medium. The results revealed that the living biomass of the cultures were more efficient to absorb Cd. The formulation of yeast, peptone medium fortified the cultures by ingredients favored the best growth yields that have the highest Cd absorption. The absorption of Cd by Pseudomonas putida was considerably influenced by the pH value of the absorption medium, contact time, biomass levels and Cd concentration. Thus, 93% of Cd was absorbed in medium containing 10 mg Cd.

#### **KEYWORDS**

Bioremediation, Cadmium, Biosorption, Biomass, *Pseudomonas putida*.

#### INTRODUCTION

A broad attention has been kept on management of environmental pollution and its control for last two decades due to hazardous materials such heavy metals. During this period, removal of heavy metals from the soil and water around industrial plants has been a challenge. In those toxic heavy metals, mercury, lead and cadmium are in the limelight due to their major impact on the environment [1]. Cadmium is the most toxic and common among the heavy metal pollutants of industrial effluents. Cd is introduced into the bodies of water from smelting, metal plating, cadmium-nickel batteries, phosphate

fertilizer, mining, pigments, stabilizers, allay industries and sewage sludge. Discharges containing cadmium, in particular, are strictly controlled due to the highly toxic nature of this element and its tendency to accumulate in the tissues of living organisms [2]. The harmful effects of cadmium include a number of acute and chronic disorders, such as *itai-itai* disease, renal damage, emphysema, hypertension, and testicular atrophy [3]. Conventional techniques for the removal of heavy metals

from wastewater, such as chemical precipitation, ion exchange, activated carbon adsorption and separation processes have limitations and become inefficient and expensive especially when the heavy metal concentration is less than 100 ppm[4]. When Cd is entered into soil, it can be taken up by plants and microbes and then enter into the food chain [5, 6, 7, 8, 9]. The use of microorganism or its biomass or its products for the recovery of metals from waste streams [10, 11]. Bioremediation is the cost effective and environmental friendly process, which has been widely accepted by microbial metal reabsorption [12]. That's why there is a need to search such metal tolerant, metal absorbent for biodegradation process. Bacterial genes determine transport systems for maintaining equilibrium intracellular concentrations balancing needs and toxicity, and for elimination of purely toxic elements such as Hg, Pb, As, Cr, Cd, and Ag [13, 14, 15]. Pseudomonas can survive in different habitats, including water, soil, and plants [16, 17]. However, microorganisms absorb the heavy metal or transform it through enzymatic reactions and cause its binding to metabolites or other chemicals in the soil. Several authors consider microorganisms as natural "biosorbants" [18, 19, 20]. Microorganisms indigenous to heavy metal-containing environments have evolved several distinct mechanisms of heavy metal tolerance[21, 22]. In this study, we attempted to determine the absorption ability of Pseudomonas putida to remove Cd from a liquid medium.

#### **MATERIALS & METHODS**

Pseudomonas putida Cultures were obtained from Microbial Type Culture Collection. In this study, the cultures were grown in water soluble medium containing Beef extract, Yeast extract, Peptone, and NaCl. The salts of Cadmium (Cadmium Sulphate) were added to the medium at concentrations of 0mg, 10mg, 20mg, 30mg, and 40mg per litre respectively. The pH of the medium was adjusted to 7.0 to 7.2 and sterilized in an autoclave at 15lbs pressure for 15min. P. putida Cultures were inoculated into the medium. Then about 2ml of the culture of the medium was incubated at 25°c for 7days. The medium with different concentrations of Cd and bacteria were kept for treatment and medium without Cadmium salts and organism served as control. Then the tubes were analyzed for the growth of organism at different concentrations of Cd, absorption of metal by organism and concentrations of metal present after treatment in an atomic absorption spectrophotometer. Values and observations were recorded.

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#### RESULT AND DISCUSSION

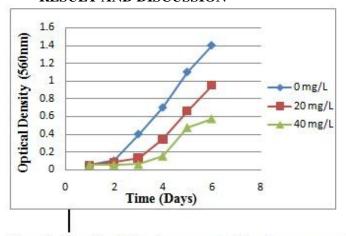


Figure 1: Growth of *Pseudomonas putida* in the presence of Cadmium

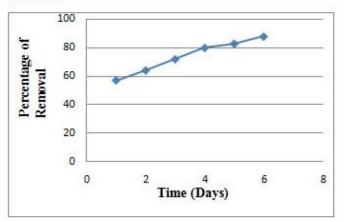


Figure 2: Cd absorption capacity of P. putida

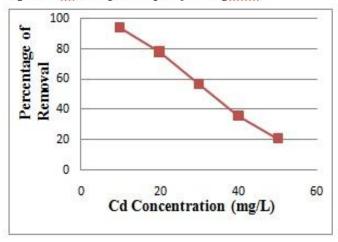


Figure 3 Cd uptake by P. putida at different concentrations

Then toxic effects of tested Cadmium on *Pseudomonas* were studied. P. *putida* resistance was investigated to the Cadmium. Live cells were cultured in solutions containing metal ion. Growth curve of P. *putida* in the presence of various concentrations of Cd is represented in

Figure 1. Results have demonstrated that the inhibitory effects of Cd increased as the concentration of this element was elevated in the culture medium. As stated in Figure 1, 40 mg/l Cd caused death in many of the bacterial cells on the first day. During the rapid growth, the ions are absorbed at the surface of microorganism. An effect of the amount of P. putida on the absorption rate of tested metals is demonstrated in Figure 2. Utilized biomass concentration of bacteria is varied between 10<sup>5</sup> and  $5 \times 10^5$  cell/ml. Results have demonstrated that the amount of bacteria saliently affected the metal absorption rate. As the bacterial biomass increases the heavy metalions absorbs rate increases too. Impact of starting concentration of Cd ions upon bioremediation is visualized in Figure 3. Initial concentrations of Cd ions highly affect the absorption rate in equilibrium phase. It is identified that at the concentration of 10 mg/l Cd ions, the ion bioremediation rate increases higher than 90%. But at more than 30 mg/l concentration of all ions, uptake rate decreases. This kind of drop might be caused by the saturation of absorption sites of the microorganism Bioremediation (Figure 3). technology microorganisms to reduce and eliminate contaminants present in soils, sediments, water, and air [23]. Bioremediation can be a cost efficient and reliable method for removing hazardous waste from the heavily contaminated sites [23, 24, 25]. A key factor to the remediation of metals is that metals are nonbiodegradable, but can be transformed through sorption, methylation, and complexation, and changes in valence state. These transformations affect the mobility and bioavailability of metals. At low concentrations, metals can serve as important components in life processes, often serving important functions in enzyme productivity. However, above certain threshold concentrations, metals can become toxic to many species [26]. Fortunately, microorganisms can affect the reactivity and mobility of metals. Microorganisms that affect the reactivity and mobility of metals can be used to detoxify heavy metals and prevent further metal contamination (Al-Homaidan, 2006; Vilensky et al., 2002).

#### CONCLUSION

P. putida was the most tolerant to Cd level up to 50 mg/L medium. The bacterial culture was more efficient to absorb Cd. The absorption of Cd by P. putida was considerably influenced by the composition of the growth medium. The absorption of Cd by P. putida was considerably influenced by its treatment, pH value of the absorption medium, contact time, biomass level and Cd concentration under these conditions 93% of Cd was absorbed. In summary, the study establishes the role and efficiency of P. putida in the absorption, accumulation, degradation and detoxification of Cadmium from environment.

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