Microstructural evolution under tempering heat treatment in AISI H13 hot-work tool steel

Prudente W. R.¹, Jefferson Fabrício C. Lins¹, Siqueira R. P.¹, Priscila S. N. Mendes¹, Rodrigo E. Pereira¹
¹ Department of Metallurgical Engineering, Fluminense Federal University, Volta Redonda, RJ, Brazil.

ABSTRACT
The steel studied in this work belongs to the hot working class of tool steels in its applications it is essential that the tool steel have high mechanical strength properties at high temperatures in order to avoid failure due to thermal fatigue, plastic deformation, crack propagation and wear. In high temperature applications the most desirable properties are the hot hardness (Red-Hardness), tempering resistance and fatigue resistance. To meet these requirement alloying elements with strong carbide formation such as Cr, V and Mo are used. The softening resistance of these tool steels is determined by changes suffered by the alloy carbides at high temperature and due to the recovery of the martensitic structure. Thus, the final effects of heat treatment on the microstructure and mechanical properties of the AISI H13 tool steel were studied. Microstructural characterization of samples was performed with the aid of the x-ray diffraction and scanning electron microscopy techniques.

Keywords: AISI H13, Alloy Carbides, Heat Treatment, Secondary Hardening, Tool Steel.

I. INTRODUCTION
Previous research on hot work tool steels has shown that softening during tempering and fatigue at high temperature is strongly linked to the microstructure and its stability at high temperatures [1]. Typical microstructures of tool steels for hot work consist of annealed martensite with high density of dislocations and alloy carbide. The softening resistance of these tool steels is determined by changes in the alloy carbides at an elevated temperature and due to the recovery of the martensitic structure.

The AISI H13 is a chromium hot-work tool steel. Its chemistry is designed to withstand the temperature, pressure, abrasion, and thermal cycling associated with various hot working operations, including plastic injection molding, die casting, forging, and extrusion. The steel has low carbon content, around 0.4 wt.%, to promote toughness. Medium chromium content, around 5 wt.% to provide good resistance to high temperature softening, small percentage of Si to improve high temperature oxidation resistance, and small molybdenum and vanadium additions (about 1%) that form stable carbides to increase resistance to erosive wear [2].

The volumetric fraction, size, morphology and distribution of the carbides have strong effects on the mechanical properties of the steel. Studies have shown that the effect of increased resistance caused by the precipitation of carbides can significantly increase the mechanical properties of the steel [3], thus the final effects of the heat treatment on the microstructure and mechanical properties of AISI H13 steel have been extensively studied.

II. EXPERIMENTAL PROCEDURE
II.1 MATERIALS
The material used in the proposed work are samples taken from an AISI H13 tool steel ingot, supplied by Villares siderurgy, the samples have dimensions of 10x10mm and 25mm length and their chemical composition is shown in Table 1.

Table 1 - Chemical composition of H13 hot-work tool steel used (wt.%).

<table>
<thead>
<tr>
<th>%C</th>
<th>%Si</th>
<th>%Mn</th>
<th>%Cr</th>
<th>%Mo</th>
<th>%V</th>
</tr>
</thead>
<tbody>
<tr>
<td>0,40</td>
<td>0,50</td>
<td>0,20</td>
<td>5,20</td>
<td>1,30</td>
<td>0,80</td>
</tr>
</tbody>
</table>

II.2 METHODS
To study the influence of tempering temperature 5 samples were austenitized at 1020°C for 0.5h and then 4 were tempered by varying the temperature parameter. The treatments are shown in Table 2.

Table 2 - Tempering conditions for hardness evolution measurements, carbides and martensite laths analysis.

<table>
<thead>
<tr>
<th>Tempering Sample</th>
<th>Temperature</th>
<th>Time</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>500°C</td>
<td>2h</td>
</tr>
<tr>
<td>2</td>
<td>550°C</td>
<td>2h</td>
</tr>
<tr>
<td>3</td>
<td>600°C</td>
<td>2h</td>
</tr>
<tr>
<td>4</td>
<td>650°C</td>
<td>2h</td>
</tr>
</tbody>
</table>
II.2.1 METALLOGRAPHIC PREPARATION

The samples were grinded on the Arotec Aropol 2V equipment, with a silica carbide sandpaper sequence of 400 #, 600, 800 #, 1000 #, 1200 #, # 1500 and 2500 #. For the polishing of the samples, 3 and 1 μm granulation diamond paste were used, and the final polishing was done with colloidal silica OPS. Finally, the etch was done with 5% nital.

II.2.2 SCANNING ELECTRON MICROSCOPY (SEM)

In this technique a beam of electrons of controlled diameter is projected on the surface of the sample to be analyzed acquiring signals caused by collisions of these electrons with the electrons of the surface of the sample.

The microstructures of the samples were observed with the aid of a microscope Zeiss EVO Scanner MA10 with LaB₆ filament.

II.2.3 X-RAY DIFFRACTION (DRX)

This technique allows us to identify the crystal structure, crystallographic texture and orientation relationship between the phases. X-ray diffraction (XRD) analysis was performed using a Shimadzu Lab X XRD-600 diffractometer, using Cu-Kα radiation, with a standard goniometer. A voltage of 30 kV, current of 30 mA was applied during a scan between 10° and 85° of 2θ, with the step speed of 2° and a scan speed of 2°/minute. The peaks were identified by comparing the experimental data and the files from the Pearson Crystal Data (PCD) using the Powder Cell program.

II.2.4 HARDNESS ROCKWELL C (HRC)

After microstructural analysis the samples were submitted to fifteen Rockwell C hardness tests with a load of 150kgf for 20s in a Süssen - Wolpert durometer Testor HT model.

III. RESULTS AND DISCUSSION

III.1 MICROSTRUCTURAL CHARACTERIZATION

The microstructure evolution of the steel was investigated by XRD, scanning electron microscopy (SEM) and the quantitative image analysis was carried out with the software IMAGEJ and the ASTM E1245 standard.

In the as-quenched condition, the parent austenite phase transforms into martensite which is accompanied by homogeneous elastic lattice deformation and by a significant increase in dislocation density [4].

Tempering promotes a diffusion type phase transformation from a quenched martensite (B.C.T. - body centered tetragonal) to a tempered martensitic structure (B.C.C. – body centered cubic). The carbides formation and growth is strongly related to tempering time and temperature. Hence, establishing a relation between tempering conditions and microstructures is of great interest [5]. The microstructure of the steel in the as quenched state is shown in Fig. 1.

![Figure 1](https://example.com/figure1.png)

**Figure 1** – AISI H13 tool steel SEM-SE micrograph of the as-quenched microstructure.

In this state the steel has a martensitic structure with high defect density, non-dissolved carbides, usually vanadium rich carbides, during austenitization (as can be observed in Fig. 3 to dissolve all carbides a high temperature is needed) and possibly some retained austenite [6].

In Fig. 2 it is possible to see the fine vanadium carbides which did not dissolve during austenitization.

![Figure 2](https://example.com/figure2.png)

**Figure 2** - AISI H13 tool steel SEM-SE micrograph of the as-quenched microstructure.

The curve shown in Fig. 3 was obtained using the Thermo-Calc software version 4.0 with the TCFe7 database. A legend with the phases found by the software is presented in table 3.
Table 3 - Legend of the Fig.3 graphic.

<table>
<thead>
<tr>
<th>Structure Type</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>BCC_A2</td>
<td>Ferrite α (CCC)</td>
</tr>
<tr>
<td>FCC_A1</td>
<td>Austenite γ (CFC)</td>
</tr>
<tr>
<td>M23C6</td>
<td>Cr rich carbide</td>
</tr>
<tr>
<td>M7C3</td>
<td>Cr rich carbide</td>
</tr>
<tr>
<td>HCP_A3</td>
<td>Mo rich carbide</td>
</tr>
<tr>
<td>FCC_A1#2</td>
<td>V rich carbide</td>
</tr>
</tbody>
</table>

Figure 3 – It is shown the phase transformation temperature as well as the alloy carbides dissolution temperature of the AISI H13 tool steel.

In Fig. 4, 5, 6 and 7 is presented the tool steel under different tempering conditions. It is possible to note how the martensite laths increase in size and thickness with increasing temperature.

Figure 4 - SEM-SE micrograph of the AISI H13 tool steel tempered at 500ºC for 2 hours SEM-SE micrograph.

Figure 5 - SEM-SE micrograph of the AISI H13 tool steel tempered at 550ºC for 2 hours SEM-SE micrograph.

Figure 6 - SEM-SE micrograph of the AISI H13 tool steel tempered at 600ºC for 2 hours.

Figure 7 - SEM-SE micrograph of the AISI H13 tool steel tempered at 650ºC for 2 hours SEM-SE micrograph.

Tempering is a very complex phenomenon originating from the as-quenched microstructure of tool steels, which consists primarily of martensite with retained austenite and carbides. When hardened steel is tempered, the tetragonality of the martensite decreases and then disappears, which results in
decrease in internal stresses. Dislocations anneal-out at 400 °C. Therefore, there is decrease in the dislocation density [4]. When steel is tempered at 600°C, the following changes in the matrix take place: internal stresses decrease; dislocations anneal-out; and coalescence of carbides [7].

The secondary hardening is a strengthening resulting of the replacement of coarse particles of Fe3C, which dissolve, by a fine dispersion of alloy carbides such as VC, Cr7C3, Cr23C6, Mo2C or Mo6C. Chromium additions result in a retardation in softening but little secondary hardening because Cr7C3 coarsens very rapidly at tempering temperatures [8]. These alloy carbides are responsible for the hardness and wear resistance in this steel.

It was noticed that with the increase of the tempering temperature, more carbides were present in the microstructure, thus, 10 images at the magnification of 5000 were made and analyzed following the ASTME1245 standard, the volumetric fraction of carbides was then calculated. The mean value is presented in table 4.

<table>
<thead>
<tr>
<th>Temperature (°C)</th>
<th>Mean (%)</th>
<th>S</th>
<th>95% CI</th>
<th>%RA</th>
</tr>
</thead>
<tbody>
<tr>
<td>500°C</td>
<td>1.4053</td>
<td>0.3049</td>
<td>0.2181</td>
<td>15.5216</td>
</tr>
<tr>
<td>550°C</td>
<td>1.6105</td>
<td>0.2697</td>
<td>0.1929</td>
<td>11.9805</td>
</tr>
<tr>
<td>600°C</td>
<td>2.0579</td>
<td>0.3003</td>
<td>0.2148</td>
<td>10.4365</td>
</tr>
<tr>
<td>650°C</td>
<td>3.7053</td>
<td>0.4433</td>
<td>0.3171</td>
<td>8.5575</td>
</tr>
</tbody>
</table>

In accordance with the ASTM standard it is shown in the table 4 the mean value of the volume fraction (p), the standard deviation (S), it is also reported the 95% confidence interval for each set of fields and the relative accuracy (%RA). It can be seen that the volume fraction highly increases with the temperature.

To analyze the martensite laths 10 images at 2000 magnification were made and analyzed through threshold and labeling, a histogram was made to understand the growth tendency of the martensite laths as shown in Fig. 8.

It can be seen that in all cases the majority of laths have 1μm width and as the temperature increases the width increases.

### III.2 X-RAY DIFFRACTION

The X-ray diffraction test was performed in all samples, but in all tests made the only phase detected was the martensite as shown in Fig. 9, it is concluded that the volume fraction of retained austenite and carbides is low, less than 4%, thus it is not detected by the test.

### III.3 HARDNESS ROCKWELL C (HRC)

Fifteen indentations were performed and the mean values found are shown in Fig.10.

The hardness is in accordance with the expected range for a material tempered for two hours. It’s evident that the secondary hardening is happening because the hardness is increasing.

### IV. CONCLUSION

The microstructural evolution under tempering of the AISI H13 tool steel was investigated describing microstructural and associate hardness evolutions. The volume fraction of carbides, could not be estimated by X-ray diffraction due to its low percentage but the results indicate that
it increases as the temperature increases. It’s also reported in the literature that the size increases with time and temperature but the particle radius was not measured in this work. Regarding the martensite lath, although the effect is not so pronounced, the martensite lath width increases when the temperature increases.

REFERENCES


